

An Efficient Approach to Chiral Cis-3,4-Epoxy Alcohols

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Abstract: A new method for the synthesis of cis-3,4-epoxy alcohols has been developed by employing hydroboration of chiral cis-vinylepoxides or syn-chlorohydrins. © 1998 Elsevier Science Ltd. All rights reserved.

Homoallylic epoxyalcohols are available from vanadium-catalyzed epoxidation of homoallylic alcohols, multi-step conversions of 2,3-epoxy halides derived from the Sharpless asymmetric epoxidation² and epoxidation of homoallylic alcohols by Cardillo's route. Sharpless asymmetric epoxidation of homoallylic alcohols is the most practical route to chiral homoallylic epoxyalcohols but this process gives only moderate enantiopurity. We report a novel and efficient approach to chiral cis-3,4-epoxy alcohols based on hydroboration of enantioenriched cis-vinylepoxides or syn-chlorohydrins.

We have recently shown that (Z)- $(\gamma$ -chloroallyl)diisopinocamphenylboranes 1a, 1b (Scheme 1) react with aldehydes to give chiral syn-vinylchlorohydrins and cis-vinylepoxides in excellent diastereo-and enantioselectivities ($\geq 95\%$ de, 90-99% ee).⁵

Table 1. Hydroboration of cis-vinylepoxides a

entry	vinylepoxide	borane ^b	3,4-epoxy alcohol ^c	yield(%) ^d
1	n-C ₁₀ H ₂₁	9-BBN	n-C ₁₀ H ₂₁ H OH	0
2	C-C ₆ H ₁₁ C C H	9-BBN	c-C ₆ H ₁₁ OH	0
3	n-C ₁₀ H ₂₁	DCHB	л-C ₁₀ H ₂₁ ОН	67
4	n-C ₅ H ₁₁	DCHB	n-C ₅ H ₁₁ OH	71
5	H O H	DCHB	Ph OH	44
6	Ph. H. O. H	DCHB	Ph. H. OH	41
7	NHBoc	DCHB	H OH NHBoc	58

^a Hydroboration was carried out in THF at rt and DCHB was prepared according to reference 7. ^b 1 Equiv of borane used. ^c 3 Equiv of sodium perborate used for oxidation. ^d Isolated yield.

Hydroboration of enantioenriched cis-3,4-epoxy-1-tetradecene or cis-1-cyclohexyl-1,2-epoxy-3-butene with 9-BBN followed by oxidation with various oxidation reagents did not yield expected cis-3,4-epoxy alcohols (Table 1, entry 1-2). Polar products similar to those expected from epoxide opening were detected.⁶ When hydroboration was conducted in THF using dicyclohexyborane (DCHB) reaction was completed in 4 hr. Use of solvent such as Et₂O, pentane or toluene resulted in much slower reaction. Choice of oxidizing reagents was crucial for the transformation of organoborane to corresponding 3,4-epoxy alcohols. Sodium perborate was the reagent of choice providing cis-3,4-epoxy alcohols in 41-71% yields (Table 1, entry 3-7).⁸

Hydrobration of acetyl protected syn-chlorohydrins provides a more efficient route to cis-3,4-epoxy alcohols (Table 2). The one pot process involved hydroboration of syn-chlorohydrins

in THF, followed by oxidation using 3 equiv of NaBO3 overnight, then the mixture was diluted with methanol (5 ml/mmol of chlorohydrin) and treated with potassium carbonate in another 4-6 hr to give expected alcohols in 65-85% yield (eq 1).9 Hydroboration with 9-BBN was sluggish compared to DCHB. Increasing the amount of 9-BBN did not increase yield (Table 2, entry 3, 4).

$$\begin{array}{c|c}
OAC \\
R & i. HBR'_{2} \\
\hline
ii. NaBO_{3}
\end{array}$$

$$\begin{array}{c|c}
OAC \\
R & i. HBR'_{2} \\
\hline
CI OH
\end{array}$$

$$\begin{array}{c|c}
K_{2}CO_{3} / MeOH \\
\hline
4 \sim 6 \text{ hr}
\end{array}$$

$$\begin{array}{c|c}
H & O \\
\hline
OH
\end{array}$$

$$\begin{array}{c|c}
H & O \\
OH
\end{array}$$

$$\begin{array}{c|c}
H & O \\
OH
\end{array}$$

Table 2. Hydroboration of syn-chlorohydrins^a

entry	chlorohydrin R	borane	chlorohydrin/borane ratio (mol/mol)	time (h)	yield (%) ^b
1	<i>n</i> -C ₁₀ H ₂₁	9-BBN	1	24	34 (61) ^c
2	<i>n</i> -C ₁₀ H ₂₁	9-BBN	1	48	37(55) ^c
3	n-C ₁₀ H ₂₁	9-BBN	1.5	24	40(57) ^{c,d}
4	n-C ₁₀ H ₂₁	9-BBN	2	24	20(37) ^{c,e}
5	n-C ₁₀ H ₂₁	DCHB	1	4	85
6	<i>i</i> -Bu	DCHB	1	4	82
7	Ph	DCHB	1	4	65
8	NHBoc	DCHB	1	4	77

^a Hydroboration was carried out in THF at rt and DCHB was freshly prepared according to reference 7.

^b Isolated yield unless noted. ^c Yield in parenthesis was based on recovered starting material. ^d 4.5 Equiv of NaBO₃ used. ^e 6 Equiv of NaBO₃ used.

Regioselectivity of hydroboration of vinylepoxides or chlorohydrins was sensitive to the purity of dicyclohexyborane. With the purified DCHB excellent regioselectivity was observed.

In summary, hydroboration of *cis*-vinylepoxides and *syn*-chlorohydrins provide access to *cis*-3,4-epoxy alcohols.

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References and Notes

- ¥ Present address: SynPhar Laboratories, Inc., 4290-91A Street, Edmonton, AB, Canada T6E 5V2.
- [1] Mihelich, E. D.; Daniels, K.; Eickhoff, D. J. J. Am. Chem. Soc. 1981, 103, 7690-7692.
- [2] Tius, M. A.; Fauq, A. J. Am. Chem. Soc. 1986, 108, 6389-6391.

Representative experimental procedure:

- [3] Lipshutz, B. H.; Barton, J. C. J. Org. Chem. 1988, 53, 4495-4499.
- [4] Rossiter, B. E.; Sharpless, K. B. J. Org. Chem. 1984, 49, 3707-3711.
- [5] a) Hu, S.; Jayaraman, S.; Oehlschlager, A. C. J. Org. Chem. 1996, 61, 7513-7520. b) Jayaraman, S.;
 Hu, S.; Oehlschlager, A. C. Tetrahedron Lett. 1995, 27, 4765-4768.
- [6] Brown, H. C.; Vara Prasad, J. V. N. J. Org. Chem. 1985, 50, 3002-3005.
- [7] Brown, H. C.; Kramer, G. W.; Levy, A. B.; Midland, M. M. Organic Synthesis via Boranes; Wiley-Interscience: New York, 1975.
- Hydroboration of cis-vinylepoxides: cis-(3R, 4S)-3,4-epoxy-1-tetradecene 7 in THF (420 mg in 1 mL THF) was introduced via syringe into 2 mmol of dicyclohexyborane in 2 mL of THF with stirring. Dicyclohexylborane was prepared from freshly distilled cyclohexene and BH3•SMe2 and further purified by sublimation. The reaction mixture was stirred at rt for 4 hr. and 910 mg of NaBO3•H2O and 0.5 mL of H2O were added. The reaction mixture was stirred overnight then 4 g of K2CO3 was added and after 2 hr the mixture was diluted with 15 mL of anhyd. Et2O. Solid was removed by filtration and washed with Et2O to give a filtrate which was concentrated under vacuum. Purification by flash chromatography using (hexane: Et2O,7:3) as the eluant gave 291 mg of cis-(3S,4S)-3,4-epoxy-tetradecan-1-ol, 8,
 - mp 33-37.5 °C. Twice recrystallization from 1% Et₂O in hexane gave mp 37.7-38.5 °C. $[\alpha]^{23}D 8.34$ (c = 2.12, Et₂O); ¹³C NMR (CDCl₃, ppm) 60.81, 56.68, 54.94, 31.88, 30.68, 29.66, 29.56, 29.53, 29.49, 29.28, 27.96, 26.46, 22.63, 14.01. ¹H NMR (CDCl₃, ppm) 3.86 (m, 2H), 3.09 (dt, J = 8.0, 4.4 Hz, 1H), 2.93 (td, J = 5.6, 4.4 Hz, 1H), 1.92-1.26 (m, 21H), 0.87 (t, J = 6.8 Hz, 3H). CIMS m/z (isobutane, relintensity) 230 [M++1(100)] 211 [(M+-18)+1(05)] April Colod for Co. Has Oct. C. 73.63; H 12.36
 - intensity) 229 [M⁺+1(100)], 211 [(M ⁺ -18)+ 1(95)]. Anal. Calcd for $C_{14}H_{28}O_2$: C, 73.63; H, 12.36. Found: C, 73.68; H, 12.44.
- [9] Hydroboration of syn-chlorohydrins: A solution of 576 mg of syn-(3S,4S)-4-acetoxy-3-chloro-1-tetradecene in 1 mL THF was introduced into 2 mmol of dicyclohexyborane THF solution (2 mL) via syringe. Stirring was continued for 4 hr at room temperature. The reaction was quenched after 4 hr at rt by addition of 920 mg of NaBO3•H₂O and 0.5 mL of H₂O. The mixture was stirred overnight then 5 mL of MeOH was added and K₂CO₃ (560 mg) were added sequentially. After 6 hr, the mixture was diluted with 15 mL of anhyd. Et₂O and 15 mL of water. The organic layer was separated and aqueous layer was extracted with Et₂O (2 X 10 mL). The combined extracts was dried over anhyd. Na₂SO₄ and then concentrated in vacuum, purified by flash column chromatography (hexane: Et₂O, 7:3) and recrystallized as above to yield 387 mg of cis-(3R,4R)-3,4-epoxy-tetradecan-1-ol, mp 38-38.5 °C. [α]²³D 8.36 (c = 2.14, Et₂O).